2011 Vol. 13, No. 10 2782–2785

ORGANIC **LETTERS**

Three-Component Reactions of Sulfonylimidates, Silyl Glyoxylates and N-tert-Butanesulfinyl Aldimines: An Efficient, Diastereoselective, and Enantioselective Synthesis of Cyclic N-Sulfonylamidines

Ming Yao and Chong-Dao Lu*

Xinjiang Technical Institute of Physics & Chemistry, Chinese Academy of Sciences, Urumqi 830011, China and Graduate University of the Chinese Academy of Sciences

clu@ms.xjb.ac.cn

Received April 18, 2011

ABSTRACT

Three-component coupling reactions of sulfonylimidates, silyl glyoxylates and N-tert-butanesulfinyl aldimines efficiently provide cyclic N-sulfonylamidines containing free endocyclic N-H. The formation of two C-C bonds (contiguous stereogenic carbons), one O-Si bond, and one $C-N$ bond, together with the cleavage of the chiral auxiliary (tert-butanesulfinyl group), occurs with excellent chemoselectivity, diastereoselectivity, and enantioselectivity in this one-pot cascade transformation.

Amidines have been identified as an important class of compounds with applications in medicinal chemistry,¹ coordination chemistry,² and supramolecular chemistry³

(1) (a) Boyd, G. V. In The Chemistry of Amidines and Imidates; Patai, S., Rappoport, Z., Eds.; Wiley: New York, 1991; Vol. 2, Chapter 8. (b) Greenhill, J. V.; Lue, P. Prog. Med. Chem. 1993, 30, 203.

(3) Katagiri, H.; Tanaka, Y.; Furusho, Y.; Yashima, E. Angew. Chem., Int. Ed. 2007, 46, 2435 and references therein.

(4) (a) King, C. J. Org. Chem. 1960, 25, 352. (b) Heck,M.-P.; Vincent, S.-P.; Murray, B. W.; Bellamy, F.; Wong, C.-H.; Mioskowski, C. J. Am. Chem. Soc. 2004, 126, 1971. (c) Kumagai, N.; Matsunaga, S.; Shibasaki, M. Angew. Chem., Int. Ed. 2004, 43, 478. (d) Nazarenko, K. G.; Shirokaya, T. I.; Tolmachev, A. A. Chem. Heterocycl. Compd. 2005, 41, 195. (e) Takuwa, T.; Minowa, T.; Onishi, J. Y.; Mukaiyama, T. Bull. Chem. Soc. Jpn. 2004, 77, 1717. (f) Lange, U. E. W.; Schafer, B.; Baucke, D.; Buschmann, E.; Mack, H. Tetrahedron Lett. 1999, 40, 7067. (g) Wang, J.; He, Z.; Chen, X.; Song, W.; Lu, P.; Wang, Y. Tetrahedron 2010, 66, 1208 and references therein.

(5) (a) Bae, I.; Han, H.; Chang, S. J. Am. Chem. Soc. 2005, 127, 2038. (b) Chang, S.; Lee, M.; Jung, D. Y.; Yoo, E. J.; Han, S. K. *J. Am. Chem.*
Soc. **2006**, 128, 12366. (c) Yoo, E. J.; Chang, S. Curr. Org. Chem. **2009**, 13, 1766. (d) Xu, X.; Li, X.; Ma, L.; Ye, N.; Weng, B. J. Am. Chem. Soc.
2008, 130, 14048. (e) Xu, X.; Ge, Z.; Cheng, D.; Ma, L.; Lu, C.; Zhang, Q.; Yao, N.; Li, X. Org. Lett. 2010, 12, 897. (f) Wang, S.; Wang, Z.; Zheng, X. Chem. Commun. 2009, 7372.

10.1021/ol201029b C 2011 American Chemical Society Published on Web 04/26/2011

due to their structural characteristics. Traditional protocols for the synthesis of the amidines rely on transforming the functional groups of amides, nitriles, aldoximes, and amines, among others.⁴ In addition, several methods involving convergent direct couplings of sulfonyl azides,⁵ isocyanides, $6 \text{ or carbodimides}^7$ with the suitable coupling partners have been developed in recent years, allowing easy access to several types of amidines. However, the applicability of these methods is restricted by intrinsic limitations in the reaction scope and generality.⁸ Meeting the demand for the efficient construction of the structurally diverse group of amidines necessitates further synthetic explorations toward efficient methods for their synthesis.

⁽²⁾ Barker, J.; Kilner, M. Coord. Chem. Rev. 1994, 133, 219.

⁽⁶⁾ Saluste, C. G.; Whitby, R. J.; Furber, M. Angew. Chem., Int. Ed. 2004, 43, 478.

⁽⁷⁾ Yu, R. T.; Rovis, T. J. Am. Chem. Soc. 2008, 130, 1833.

⁽⁸⁾ For example, only cyclic N-sulfonylamidines with a protected nitrogen at the N -amino position can be accessed by the direct-coupling methods in refs 5b and 5e. The analogues provided in refs 6 and 7 are N-alkyl amidines.

Herein, we report a three-component coupling reaction of sulfonylimidates, silyl glyoxylates 9 and N-tert-butanesulfinyl aldimines for an efficient, convergent, asymmetric synthesis of substituted cyclic N-sulfonylamidines possessing free N-H at the cyclic nitrogen atom (Scheme 1). In this reaction, the nucleophilic addition of lithium aza-enolates Li-1 to silyl glyoxylates 2 triggers a Brook rearrangement,¹⁰ and the generated enolates 5 undergo addition to (R_s) -N-tert-butanesulfinyl aldimines 3, in which the Ellman imines¹¹ participate as the second electrophile. The anionic nitrogen-induced cyclization and subsequent desulfinylation by nucleophilic attack by the extruding ethoxide give enantioenriched cyclic N-sulfonylamidines 4. To the best of our knowledge, this is the first report that explores the use of lithium aza-enolates as nucleophiles to initiate a Brook rearrangement and the first report that azomethines serve as the second electrophile in silyl glyoxylate-mediated cascades.¹²

Scheme 1. Silyl Glyoxylates-Mediated Cascade Reaction for the Synthesis of Chiral Cyclic N-Sulfonylamidines

We began our investigation with the coupling reaction of 1a (Ar = 4-MeC₆H₄), 2a (R¹ = tBu), and 3a (R² = Ph) in the presence of several readily available metal amides.¹³ After some pilot studies, the optimized reaction conditions were found to be the metalation of sulfonylimidate 1a with LHMDS at -78 °C, followed by the sequential addition of (R_S) -tert-butanesulfinyl aldimine 3a and silyl glyoxylate

2a. The reaction mixture was stirred at -78 °C for 5 h and warmed gradually to -5 °C. The three-component product 4a was obtained in 82% isolated yield with excellent diastereoselectivity and enantioselectivity (Table 1, entry 1).

 a Imine 3 and silyl glyoxylate 2 were added sequentially to a 0.1 M enolate solution in THF at -78 °C (see the Supporting Information for the detailed procedures). The diastereoselectivity (dr) was determined from ¹H NMR spectra of crude reaction mixtures. ^b Isolated yields after silica gel chromatography. ^c Determined by HPLC with a Chiralcel OD-H column. ^d Using (S_S) -3a as a substrate in place of (R_S) -3a. e Amidines of *tert*-butanesulfinyl-4n and 4n were obtained in 63 and 13% yields, respectively. ^f Substrate ratio: sulfonylimidate 1 (2.5 equiv), silylglyoxylate 2 (2.5 equiv), imine 3 (1.0 equiv). ^{*s*} Substrate ratio: sulfonylimidate 1 (2.0 equiv), silylglyoxylate 2 (2.0 equiv), imine 3 (1.0 equiv) . ^h One-gram scale reaction.

This high yield suggested that the lithium enamide of sulfonylimidate 1a reacts as a discriminating nucleophile with a strong preference for the silylglyoxylate 2a over N-tert-butanesulfinyl aldimine 3a (Scheme 2). To verify this, several control experiments were conducted. The aldimine 3a was added to the solution of the lithium enamide of sulfonylimidate 1a at -78 °C, and no product was observed. Even when the reaction mixture was warmed to room temperature, only the starting materials were recovered (eq 2). Furthermore, when N-Ts phenyl aldimine 6 was used instead of the sulfinyl aldimine 3a as an electrophile under the standard three-component coupling conditions, the major product appeared to be the twocomponent adduct 7 derived from the sulfonylimidate 1a and the N-Ts imine 6 (eq 3).¹⁴ Studies of the nucleophilic

⁽⁹⁾ The synthetic applications of silyl glyoxylates are growing in prominence. For related examples see: (a) Nicewicz, D. A.; Johnson, J. S. J. Am. Chem. Soc. 2005, 127, 6170. (b) Nicewicz, D. A.; Satterfield, A. D.; Schmitt, D. S.; Johnson, J. S. J. Am. Chem. Soc. 2008, 130, 17281. (c) Greszler, S. N.; Johnson, J. S. Org. Lett. 2009, 11, 827. (d) Greszler, S. N.; Johnson, J. S. Angew. Chem., Int. Ed. 2009, 48, 3689. (e) Greszler, S. N.; Malinowski, J. T.; Johnson, J. S. J. Am. Chem. Soc. 2010, 132, 17393. (f) Boyce, G. R.; Johnson, J. S. Angew. Chem., Int. Ed. 2010, 49, 8930. (g) Schmitt, D. C.; Johnson, J. S. Org. Lett. 2010, 12, 944.

^{(10) (}a) Brook, A. G. Acc. Chem. Res. 1974, 7, 77. (b) Bulman Page, P. C.; Klair, S. S.; Rosenthal, S. Chem. Soc. Rev. 1990, 19, 147. (c) Moser, W. H. Tetrahedron 2001, 57, 2065.

⁽¹¹⁾ For reviews of tert-butanesulfinamide: (a) Ferreira, F.; Botuha, C.; Chemla, F.; Perez-Luna, A. Chem. Soc. Rev. 2009, 38, 1162. (b) Robak, M. T.; Herbage, M. A.; Ellman, J. A. Chem. Rev. 2010, 110, 3600.

⁽¹²⁾ For examples of aryl acylsilanes acting as conjunctive reagents for the union of lithium amide enolates and N-diphenylphosphinyl imines, see: (a) Lettan, R. B.; Woodward, C. C.; Scheidt, K. A. Angew. Chem., Int. Ed. 2008, 47, 2294. (b) Lettan, R. B.; Galliford, C. V.; Woodward, C. C.; Scheidt, K. A. J. Am. Chem. Soc. 2010, 132, 8805. It is noteworthy that the aryl acylsilanes failed to enable the three-component coupling reaction described here; see ref 17. (13) See the Supporting Information for details.

Scheme 2. Control Experiments

behaviors of lithium ester enolates (Li-8a and Li-8b) toward the mixture of 2a and 3a indicated that the predominant reaction pathway is addition to sulfinylimine 3a, rather than the addition to the silylglyoxylate 2a (eq 4). These results reveal that the three-component reaction described here exhibits unusual and exquisite chemoselectivity.

The scope and limitations of this method were investigated, and are summarized in Table 1. The reactions provided the desired products in good isolated yields $(59-91\%)$ with excellent diastereoselectivity ($>$ 20:1 dr) and enantioselectivity $(94-99.3\%$ ee). The enantiomer of the cyclic sulfonylamidine **4a** could be obtained by employing (S_s) -tertbutanesulfinyl phenyl aldimines (entry 1). The crystalline compound of the (R) -mandelic acid derivative from 4j was prepared, which allowed us to determine the absolute stereochemistry of the newly formed contiguous stereogenic carbons by single-crystal X-ray diffraction.¹⁵

The aryl aldimines bearing electron-donating groups were more reactive than those bearing electron-withdrawing groups. Higher yields were obtained, and an excess of more than 2.0 equiv of the sulfonylimidates and silylglyoxylates was unnecessary: yields of $80-91\%$ were obtained with 2.0 equiv, compared to the 72% with 2.5 equiv (Table 1, entrie $4-8$, 10, and 11 vs entries $2-3$). Good yields and excellent enantioselectivities were achieved using the following compounds: phenyl aldimines with a methoxy substituent in the *ortho*, *meta*, and *para* positions; heteroaromatic aldimines; and unbranched aliphatic aldimines (Table 1, entries $5-7$, $10-13$). Unfortunately, sulfinylimine derived from bulky pivaldehyde was inert. When the α , β -unsaturated aldimine 3n was used in the reaction (Table 1, entry 14), two products were observed: the $N-H$ sulfonylamidine 4n and the N-sulfinyl sulfonylamidine

tert-butanesulfinyl-4n. The latter compound was the product without cleavage of the sulfinyl auxiliary. Warming the reaction mixture to -5 °C resulted in an intractable mixture of products. Notably, the enantioselectivity for 4k (99.3% ee, entry 11) is consistent with the enantiopurity of (R_S) -tert-butanesulfinamide (99.3% ee) used to prepare imine 3k.¹⁶ This implies that a complete chirality transfer from the chiral auxiliary occurred in this example. Onegram scale preparation of 3k gave comparable ee and a slightly lower yield compared to a small scale process because the desulfinylation in the larger scale reaction is slower, and prolonging the reaction time led to partial decomposition (entry 11). The cyclic arylsulfonyl amidines incorporating a benzyl ester or other arylsulfonyl groups $Ar = Ph$, PMP) can be prepared using the corresponding starting materials (Table 1, entries $15-17$). Aryl acylsilanes such as PhCOTBS (10) did not function as useful reagents to enable the three-component coupling reaction because the intramolecular electrophilic trapping is more efficient.¹⁷

Scheme 3. Rationale of Stereochemistry

To rationalize the stereochemistry of this transformation, a transition state TS-1 was proposed (Scheme 3) according to Ellman's six/four-membered bicyclic transition state with a preferred chair conformation.18 It has been well established by Johnson and co-workers that (Z)-glycolate enolates were preferentially formed over (E) -isomer in the lithium enolate-initiated Brook rearrangements of silyl glyoxylates.^{9g} The intermediate glycolate enolates (Z)-5 approach the imines (E)-(R_s)-3 from the

⁽¹⁷⁾ The cyclopropane derivative 12 was predominantly formed with excellent diastereoselectivity by intramolecular nucleophilic trapping of the α -silyloxy carbanion 11 (intermediate of Brook rearrangement). The stereochemistry of the product 12 was determined from 2D ROESY spectrum. For details, see the Supporting Information. For similar reaction behavior in the addition of lithium ketone enolates to acylsilanes, see: Takeda, K.; Nakatani, J.; Nakamura, H.; Sako, K.; Yoshii, E.; Yamaguchi, K. Synlett 1993, 841.

(18) For the addition of lithium (or titanium) ester enolates to sulfinylimines rationalized by a six/four-membered bicyclic transition state, see: (a) Davis, F. A.; Reddy, R. T.; Reddy, R. E. J. Org. Chem. 1992, 57, 6387. (b) Tang, T. P.; Ellman, J. A. J. Org. Chem. 2002, 67, 7819.

⁽¹⁴⁾ For studies on the addition of sulfonylimidates to N-Ts imines in the presence of catalytic amounts of base or alkaline earth metal complexes, see: (a) Matsubara, R.; Berthiol, F.; Kobayashi, S. J. Am.
Chem. Soc. 2008, 130, 1804. (b) Nguyen, H. V.; Matsubara, R.; Kobayashi, S. Angew. Chem., Int. Ed. 2009, 48, 5927. (c) Matsubara, R.; Nguyen, H. V.; Kobayashi, S. Bull. Chem. Soc. Jpn. 2009, 82, 1083.

⁽¹⁵⁾ The absolute stereochemistry of the product 4j was determined by X-ray crystallography of the (R) -mandelic acid derivative of the enantiopure 4j, with $4a-4i$ and $4k-4q$ assigned by analogy. Single crystal structure of racemic 4q were also provided to determine the relative configuration. See the Supporting Information for details.

⁽¹⁶⁾ The enantioselectivity of tert-butanesulfinamide was determined by HPLC with a Chiralcel OD column. See the Supporting Information for details.

opposite direction of the sterically hindered tert-butyl group (the Si face of imines 3). Nucleophilic attack by the Re face of the lithium enolates 5 is facilitated and favored by the complexation of the lithium and gives the products with the observed stereochemistry.

In summary, we have devoloped a Brook rearrangement-mediated, three-component coupling reaction of sulfonylimidates, silyl glyoxylates and N-tert-butanesulfinyl aldimines for efficient asymmetric synthesis of substituted cyclic sulfonyl amidines. The cascade promotes the formation of two $C-C$ bonds (contiguous stereogenic carbons), one $C-N$ bond, and one $O-Si$ bond, together with cleavage of the chiral auxiliary (tert-butanesulfinyl group) in a one-pot operation. The reaction occurs with exquisite chemoselectivity, excellent diastereoselectivity, and excellent enantioselective induction.

Acknowledgment. This work was partially supported by the Chinese Academy of Sciences, the National Natural Science Foundation of China (20972182), and the "Western Light" program of the Chinese Academy of Sciences. We are indebted to Prof. Armen Zakarian at UCSB (U.S.A.) for helpful comments. We thank Prof. Liu-Zhu Gong's research group at USTC (China) for their help in establishing the conditions for HPLC resolution of the racemic compound 4a.

Supporting Information Available. Experimental details, characterization data, and crystal structure data of the racemic $4q$ and the (R) -mandelic acid derivative of the enantiopure 4j. This material is available free of charge via the Internet at http://pubs.acs.org.